### **Original** Article

# Complex secondary metabolites from *Ludwigia leptocarpa* with potent antibacterial and antioxidant activities

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Summary Diarrhea continues to be one of the most common causes of morbidity and mortality among infants and children in developing countries. The aim of the present study was to evaluate the antibacterial and antioxidant activities of extracts and compounds from Ludwigia leptocarpa, a plant traditionally used for its vermifugal, anti-dysenteric, and antimicrobial properties. A methanol extract was prepared by maceration of the dried plant and this was successively extracted with ethyl acetate to obtain an EtOAc extract and with n-butanol to obtain an n-BuOH extract. Column chromatography of the EtOAc and n-BuOH extracts was followed by purification of different fractions, leading to the isolation of 10 known compounds. Structures of isolated compounds were assigned on the basis of spectral analysis and by comparison to structures of compounds described in the literature. Antioxidant activity was evaluated using 1,1-diphenyl-2-picrylhydrazyl (DPPH) and gallic acid equivalent antioxidant capacity (GAEAC) assays. Antibacterial activity was assessed with the minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) with respect to strains of a Gram-positive bacterium, Staphylococcus aureus (a major cause of community and hospital-associated infection), and Gram-negative multidrug-resistant bacteria, Vibrio cholerae (a cause of cholera) and Shigella flexneri (a cause of shigellosis). All of the extracts showed different degrees of antioxidant and antibacterial activities. 2<sup>β</sup>-hydroxyoleanolic acid, (2R,3S,2''S)-3''',4',4''',5,5'',7,7''-heptahydroxy-3,8"-biflavanone, and luteolin-8-C-glucoside displayed the most potent antibacterial and antioxidant properties, and these properties were in some cases equal to or more potent than those of reference drugs. Overall, the present results show that L. leptocarpa has the potential to be a natural source of anti-diarrheal and antioxidant products, so further investigation is warranted.

Keywords: Ludwigia leptocarpa, Onagraceae, triterpenoids, flavonoids, antibacterial, antioxidant

#### 1. Introduction

In developing countries, and particularly in Africa, poor sanitation exposes people to a wider array of microbial pathogens, increasing their susceptibility to bacterial infections (1). Each year, 3 million children are reported to die of diarrheal diseases. Cholera is a leading diarrheal disease in terms of its severity and outcomes. Several epidemics of cholera have been

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reported in different parts of Cameroon and abroad (2-5). Vibrio cholerae strains belonging to the O1 and O139 serogroups cause epidemics and pandemics of cholera (6,7). Over the past few years, reported cases of cholera have increased steadily, numbering more than 300,000 cases and including more than 7,500 deaths in 2010 (8). As populations of poor countries continue to coalesce in mega-cities with low levels of sanitation and people move rapidly around the globe, new and more virulent strains of V. cholerae are expected to disseminate more rapidly (9,10). This makes cholera one of the most rapidly fatal infectious illnesses known.

The continuous emergence of multi-drug-resistant (MDR) Vibrio cholerae strains drastically reduces the efficacy of our antibiotic armory and, consequently, increases the frequency of the rapeutic failure (11, 12). In many regions affected by this pathogen, local and indigenous plants are often the only available means of treating such infections. Among the known plant species on Earth (estimated at 250,000-500,000), only a small fraction have been investigated for the presence of antimicrobial compounds and only 1-10% of plants are used by humans (13,14). Natural plant products also act as antioxidants. These include phenolic compounds, alkaloids, terpenoids, and essential oils. Plant-based antioxidant compounds (15) play a defensive role by preventing the generation of free radicals and hence are extremely beneficial to alleviating infectious diseases that generate free radicals as well as diseases caused by oxidative stress such as cardiovascular diseases, diabetes, inflammation, degenerative diseases, cancer, anemia, and ischemia (16).

Ludwigia leptocarpa (Nutt) Hara (Onagraceae or Oenotheraceae) is a herbaceous plant species that is also readily found in North America and in tropical Africa (17). In traditional medicine in Nigeria, an infusion of the plant is part of a mixture used to treat rheumatism (18). A leaf infusion has laxative, vermifugal, and antidysenteric properties. Previous studies of this genus have revealed the presence of flavonoids (19,20), cerebrosides, and triterpenoids (20,21). A study recently reported that alcoholic extracts of the leaves of L. octovalvis, L. abyssinica, and L. decurrens potentially have antioxidant, antibacterial, and antifungal activities (22,23). To the extent known, no study has reported on the antioxidant and antibacterial properties of L. leptocarpa with respect to bacterial strains causing diarrhea. Hence, the aim of this study was to investigate the antibacterial and antioxidant properties of extracts and compounds from *L. leptocarpa*.

#### 2. Materials and Methods

#### 2.1. Experimental

IR spectra were recorded with a Shimadzu FT-IR-8400S (Shimadzu, France) spectrophotometer. <sup>1</sup>H (500 MHz) and <sup>13</sup>C (125 MHz) nuclear magnetic resonance (NMR) spectra were recorded on a BRUKER Avance DRX-500 spectrometer (Bruker, Wissembourg, France) equipped with a BBFO + 5 mm probe. <sup>1</sup>H (600 MHz) and <sup>13</sup>C (150 MHz) NMR spectra were recorded on a BRUKER Avance III-600 spectrometer (Bruker, Wissembourg, France) equipped with a cryoplatform using CD<sub>3</sub>OD, with tetramethylsilane (TMS) as the internal standard. Time-of-flight electrospray ionization mass spectrometry (TOF-ESIMS) and highresolution time of flight electrospray ionization mass spectrometry (HR-TOFESIMS) experiments were performed using a Micromass Q-TOF micro instrument (Manchester, UK) with an electrospray source. The samples were introduced by direct infusion in a solution of methanol (MeOH) at a rate of 5 µL min<sup>-1</sup>. Column chromatography was performed on Merck silica gel (VWR, France) 60 (70-230 mesh) and gel permeation chromatography was performed on Sephadex LH-20 (VWR, France), while thin layer chromatography (TLC) was carried out on silica gel GF254 pre-coated plates with detection accomplished by spraying with 50% H<sub>2</sub>SO<sub>4</sub> followed by heating at 100°C or by visualization with an ultra-violet (UV) lamp at 254 and 365 nm.

#### 2.2. Plant material

*L. leptocarpa* plants were collected in the village of Foto (Menoua Division, Western region of Cameroon), in April 2011. Authentication was performed by Victor Nana, a botanist at the Cameroon National Herbarium, Yaoundé, where a voucher specimen (N° 38782/HNC) was deposited.

#### 2.3. Extraction and isolation

Dried *L. leptocarpa* (4 kg) was extracted with MeOH at room temperature for 3 days, and the extract was concentrated to dryness under reduced pressure to yield a dark crude extract (102 g). Part of the residue obtained (97 g) was suspended in water (200 mL) and successively extracted with ethyl acetate (EtOAc) and n-butanol (*n*-BuOH). The result was concentrated to dryness under reduced pressure to respectively yield EtOAc (20 g) and *n*-BuOH (40 g) extracts.

In accordance with antimicrobial and antioxidant assays, the EtOAc and *n*-BuOH extracts were submitted to further separation and purification. Part of the EtOAc extract (15 g) was purified over a silica gel column and eluted with hexane containing increasing concentrations of EtOAc (10%, 20%, 30%, 40%, 50%, 60%, 70%, and 80%). The purified extracted was also eluted with EtOAc containing increasing concentrations of MeOH (10% and 20%). Six fractions were obtained: A, B, C, D, E, and F. Fraction D (1.7 g) was purified over a silica gel column and eluted with a hexane-EtOAc mixture (7:3) to yield compounds **1** and **2** (17 mg and 22 mg,

respectively). Fraction E (3.1 g) was purified over a silica gel column and eluted with a hexane-EtOAc mixture (6:4) to yield compound 3 (17 mg). Part of the n-BuOH extract (30 g) was purified over a silica gel column and eluted with EtOAc containing increasing concentrations of MeOH (10%, 20%, 30%, 40%, and 50%). Five fractions ( $G_1$ - $G_5$ ) were obtained. Fraction  $G_1$ (2.5 g) was purified over a silica gel column and eluted with EtOAc to yield the compounds 4 (19 mg) and 5 (16 mg). Fraction  $G_2$  (3.1 g) was purified over a silica gel column and eluted with an EtOAc-MeOH mixture (8.5:1.5) to yield compounds 5 (25 mg) and 6 (13 mg). Fractions  $G_3$  and  $G_4$  (5.4 g) were combined and purified over a silica gel column; the fractions were then eluted with an EtOAc-MeOH-H2O mixture (8:1:1) to yield the compounds 7 (38 mg) and 8 (24 mg). Fraction G5 (2.5 g) was purified over a silica gel column and eluted with an EtOAc-MeOH-H<sub>2</sub>O mixture (7:2:1) to yield the compounds 9 (66 mg) and 10 (40 mg).

Oleanolic acid (1): white amorphous powder from hexane-EtOAc;  $C_{30}H_{48}O_3$ .

 $2\beta$ -hydroxyoleanolic acid (2): white amorphous powder from hexane-EtOAc;  $C_{30}H_{48}O_4$ .

(2R,3S,2"S)-3"',4',4''',5,5",7,7"-heptahydroxy-3,8"biflavanone (**3**): white amorphous powder from hexane-EtOAc; C<sub>30</sub>H<sub>22</sub>O<sub>11</sub>; high resolution electron impact mass spectrometry (HRESIMS, positive-ion mode) m/z: 581.1057 [M + Na]<sup>+</sup>(calcd. for C<sub>30</sub>H<sub>22</sub>O<sub>11</sub>Na :581.1060).

Ellagic acid (4): yellow powder from EtOAc;  $C_{14}H_6O_8$ .

 $\beta$ -sitosterol-3-*O*- $\beta$ -D-glucopyranoside (5): white amorphous powder from EtOAcC<sub>35</sub>H<sub>60</sub>O<sub>6</sub>.

Luteolin-8-C-glucoside (6): yellow amorphous powder from  $EtOAcC_{21}H_{20}O_{11}$ .

28-*O*-β-D-xylopyranosyl- $(1 \rightarrow 4)$ -α-Lrhamnopyranosyl- $(1\rightarrow 2)$ - $[\alpha$ -L-arabinopyranosyl- $(1\rightarrow 3)$ ]-4-*O*-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)β-D-fucopyranosyl zanhic acid (7): white amorphous solid from EtOAc; C<sub>60</sub>H<sub>94</sub>O<sub>27</sub>; HRESIMS (positive-ion mode) m/z: 1269.5870 [M + Na]<sup>+</sup>(calcd. for C<sub>60</sub>H<sub>94</sub>O<sub>27</sub>Na : 1269.5880).

3-*O*-β-D-glucopyranosyl-28-*O*-β-D-xylopyranosyl-(1→4)-α-L-rhamnopyranosyl-(1→2)-4-*O*-(3'hydroxybutanoyloxy-3-hydroxybutanoyloxy)-β-Dfucopyranosyl medicagenic acid (8): white amorphous solid from EtOAc; C<sub>61</sub>H<sub>96</sub>O<sub>27</sub>; HRESIMS (positive-ion mode) m/z: 1283.6044 [M + Na]<sup>+</sup> (calcd. for C<sub>61</sub>H<sub>96</sub>O<sub>27</sub>Na : 1283.6037).

3-*O*-β-D-glucopyranosyl-28-*O*-β-D-xylopyranosyl-(1→4)-α-L-rhamnopyranosyl(1→2)-[α-Larabinopyranosyl-(→3)]-4-*O*-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)-β-D-fucopyranosyl zanhic acid (**9**): white amorphous solid from EtOAc; C<sub>66</sub>H<sub>104</sub>O<sub>32</sub>; HRESIMS (positive-ion mode) m/ z: 1431.6395[M+Na]<sup>+</sup>(calcd. for C<sub>66</sub>H<sub>104</sub>O<sub>32</sub>Na : 1431.6408).

 $3 - O - \beta - D - g | u copyranosyl - (1 \rightarrow 4) - \beta - D -$ 

glucopyranosyl-28-*O*- $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 4)- $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)-[ $\alpha$ -L- arabinopyranosyl-(1 $\rightarrow$ 3)]-4-*O*-(3'-hydroxybutanoyloxy)- $\beta$ -D-fucopyranosyl zanhic acid (10): white amorphous solid from EtOAc; C<sub>72</sub>H<sub>114</sub>O<sub>37</sub>; HRESIMS (positive-ion mode) m/z: 1593.6927[M + Na]<sup>+</sup>(calcd. for C<sub>72</sub>H<sub>114</sub>O<sub>37</sub>Na : 1593.6937).

#### 2.4. Antibacterial assay

#### 2.4.1. Microorganisms

A total of six bacterial strains were tested for their susceptibility to compounds and these strains were from our laboratory collection (kindly provided by Dr. T. Ramamurthy, NICED, Kolkata). Among the clinical strains of Vibrio cholerae used in this study, strain NB2 belongs to the O1 serotype and strain SG24(1) belongs to the O139 serotype. These strains are able to produce cholera toxin and hemolysin (24,25). The other strains used in this study were non-O1 and non-O139 strains of V. cholerae (strains CO6 and PC2) (24) and strains of Shigella flexneri (26). The non-O1 and non-O139 strains of V. cholerae were positive for hemolysin production but negative for cholera toxin production (24). An American Type Culture Collection (ATCC) strain of Staphylococcus aureus, ATCC 25923, was used for quality control. The bacterial strains were maintained on an agar slant at 4°C and subcultured on appropriate fresh agar plates 24 h prior to any antibacterial testing. Mueller Hinton Agar (MHA) was used to activate bacteria. Mueller Hinton Broth (MHB) was used to determine minimum inhibitory concentrations (MICs) and nutrient agar (Hi-Media) was used to determine minimum bactericidal concentrations (MBCs).

#### 2.4.2. Determination of MICs and MBCs

MICs and MBCs of extracts/compounds were assessed using the broth microdilution method recommended by the National Committee for Clinical Laboratory Standards (27,28) with slight modifications. Each test sample was dissolved in dimethylsulfoxide (DMSO, Fisher chemicals) to yield a stock solution. Ninetysix-well round-bottom sterile plates were prepared by dispensing 180  $\mu$ L of the inoculated broth (1 × 10<sup>6</sup> CFU/ mL) into each well. A 20 µL aliquot of a compound was added. The concentration of the tested samples varied from 0.125 to 1,024  $\mu$ g/mL. The final concentration of DMSO in each well was <1% [preliminary analyses with 1% (v/v) DMSO did not inhibit the growth of the test organisms]. Dilutions of ampicillin (Sigma-Aldrich, Steinheim, Germany) and tetracycline (Sigma-Aldrich, Steinheim, Germany) served as positive controls, while broth with 20 µL of DMSO was used as a negative control. Plates were covered and incubated

for 24 h at 37°C. After incubation, MICs were read visually; bacteria were plated on nutrient agar (Conda, Madrid, Spain) and incubated at 37°C for 24 h. The lowest concentrations that yielded no growth after this subculturing served as the MBC.

#### 2.5. Antioxidant assay

#### 2.5.1. DPPH free radical scavenging assay

The free radical scavenging activity of extracts as well as their isolated compounds was assessed in accordance with the methods of Brand-Williams et al. (29) with slight modifications. Briefly, different concentrations (10 to 2,000 µg/mL) of extracts or compounds and vitamin C (Sigma-Aldrich, Steinheim, Germany) were thoroughly mixed with 3 mL of a methanolic DPPH solution (20 mg/L) in test tubes and the resulting solution was allowed to stand for 30 minutes at room temperature before the optical density (OD) was measured at 517 nm. The measurement was repeated 3 times and an average of those readings was determined. The percentage radical scavenging activity was calculated using the following formula: % scavenging [DPPH] =  $[(A_0 - A_1)/A_0] \times 100$ . Here,  $A_0$  is the absorbance of the control and  $A_1$  is the absorbance in the presence of the sample. The  $IC_{50}$  was determined from a graph obtained using standard vitamin C by using the formula "y = mx + c" for the slope of the graph.

## 2.5.2. Gallic acid equivalent antioxidant capacity (GAEAC) assay

A GAEAC assay was performed as previously described (30) with slight modifications. In a quartz cuvette, 20 µL of laccase (1 mM stock solution), 20 µL of a test sample, and 10 µL of ABTS (2,2'-azinobis(3ethylbenzothiazoline-6-sulfonic acid) (74 mM of stock solution) were added to 950 µL of an acetate buffer (pH = 5.0, 100 mM). The laccase was purified from Sclerotinia sclerotiorum according to a previously described protocol (31). The sample concentrations in the assay mixture were 800, 400, 200, 100, and 10 µg/mL for the extracts and 200, 100, 50, 25, and 125.5 µg/mL for the isolated compounds. The content of the generated ABTS<sup>•+</sup> radical was measured at 420 nm after reaction for 240 s and this measurement was converted to the gallic acid equivalent using a calibration curve (Pearson's correlation coefficient: r = 0.996) created with 0, 4, 10, 14, 28, 56, and 84  $\mu$ M of gallic acid rather than Trolox. Experiments were done in triplicate.

#### 2.6. Statistical analysis

Data were analyzed using one-way analysis of variance followed by the Waller-Duncan post-hoc test. Results are expressed as the mean  $\pm$  standard deviation (SD).

p < 0.05 was considered to indicate a significant difference. All analyses were performed using the software Statistical Package for Social Sciences (SPSS, version 12.0).

#### 3. Results and Discussion

#### 3.1. Chemical analysis

In accordance with antibacterial assays of the MeOH, EtOAc, and n-BuOH extracts, the EtOAc and n-BuOH extracts were further separated and purified. This led to the isolation of 10 compounds. Structures (Figure 1) of these compounds have been assigned on the basis of spectroscopic data (<sup>1</sup>H and <sup>13</sup>C NMR, <sup>1</sup>H-<sup>1</sup>H COSY, HSQC, HMBC, ROESY, and NOESY), mass spectra, and by comparison of those compounds to compounds described in the literature. Hence, the isolated compounds were identified as oleanolic acid (1) (32); 2β-hydroxyoleanolic acid (2) (32); (2R,3S,2"S)-3",4',4",5,5",7,7"-heptahydroxy-3,8"-biflavanone (3) (33); ellagic acid (4) (34);  $3-O-\beta$ -D-glucopyranosyl- $\beta$ -sitosterol (5) (35); luteolin-8-C-glucoside (6) (36); 28-O- $\beta$ -Dxylopyranosyl- $(1\rightarrow 4)$ - $\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 2)$ - $[\alpha$ -Larabinopyranosyl- $(1\rightarrow 3)$ ]-4-O-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)-β-D-fucopyranosyl zanhic acid (7) (21);  $3-O-\beta$ -D-glucopyranosyl-28- $O-\beta$ -Dxylopyranosyl- $(1 \rightarrow 4)$ - $\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 2)$ -4-O-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)- $\beta$ -D-fucopyranosyl medicagenic acid (8) (21); 3-O- $\beta$ -Dglucopyranosyl-28-O- $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 4)- $\alpha$ -Lrhamnopyranosyl( $1\rightarrow 2$ )-[ $\alpha$ -L-arabinopyranosyl-( $\rightarrow 3$ )]-4-O-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)- $\beta$ -D-fucopyranosyl zanhic acid (9) (20); and 3-O- $\beta$ -Dglucopyranosyl- $(1\rightarrow 4)$ - $\beta$ -D-glucopyranosyl-28-O- $\beta$ -Dxylopyranosyl- $(1\rightarrow 4)$ - $\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 2)$ - $\lceil \alpha$ -Larabinopyranosyl- $(1\rightarrow 3)$ ]-4-O-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)- $\beta$ -D-fucopyranosyl zanhic acid **(10)** (*21*).

#### 3.2. Antibacterial activity

The susceptibility pattern and inhibition parameters of the tested organisms to the extracts and isolated compounds are indicated below (Table 1). Wells containing a concentration of 64-512 µg/mL of MeOH, EtOAc, and *n*-BuOH extracts inhibited the visible growth of all bacterial species. The most sensitive bacterial species were *S. aureus* and *S. flexneri*, while *V. cholerae* SG24(1) and *V. cholerae* NB2 were the species that were most resistant to the tested samples. All 3 plant extracts displayed less antibacterial activity than tetracycline. However, these extracts were active against *V. cholerae* NB2, *V. cholerae* PC2, and *S. flexneri* which were not sensitive to ampicillin. The antimicrobial activity of a plant extract was considered to be good if its MIC was less than 100.0 µg/mL,

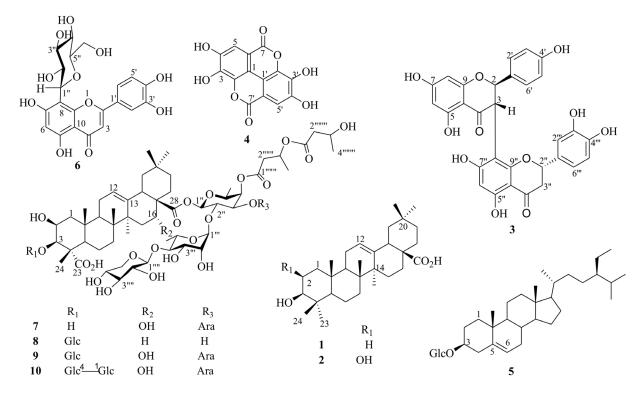


Figure 1. Structures of compounds isolated from the plant *L. leptocarpa*.

moderate if its MIC was from 100.0 to 500.0 µg/mL and poor if its MIC was over 500.0 µg/mL (37). Hence, the MeOH, EtOAc, and *n*-BuOH extracts of *L. leptocarpa* exhibited good activity against *S. aureus*, with an MIC of 64 µg/mL, whereas only the MeOH extract displayed poor activity against *V. cholerae* SG24(1). The present results for extracts of *L. leptocarpa* indicated that this plant species is a potential source of antibacterial agents. This *in vitro* study corroborated a previous study that found that alcoholic extracts of *L. octovalvis*, *L. abyssinica*, and *L. decurrens* leaves inhibited *Staphylococcus aureus* (22,23,38).

Compound 3 had the lowest MICs and MBCs, 2 µg/mL, for S. aureus; this compound has promise as an antibacterial since it was more potent at inhibiting S. aureus than the reference antibacterials ampicillin (MIC of 16 µg/mL and MBC of 16 µg/mL) and tetracycline (MIC of 16  $\mu$ g/mL and MBC of 128  $\mu$ g/mL) were. However, a MeOH extract had the highest MIC, 512 µg/mL, for V. cholerae SG24(1) while a MeOH extract had the highest MBC, 512 µg/mL, for V. cholerae SG24(1), V. cholerae CO6, and V. cholerae PC2. A lower MBC or MIC ( $\leq 4$ ) means that a minimum amount of the plant extract or isolated compound was needed to kill the bacterial species while a higher value means that a comparatively higher concentration of the extract or compound was needed to control of the microorganism (39).

Ranked in order of antibacterial activity, compound **3** isolated from *L. leptocarpa* had the most potent antibacterial activity, followed by compound 6,

compound 2, compound 4, compounds 8 and 9, compound 10, compound 7, and then compound 1. Compounds 3, 6, 2, 4, 8, 9, and 10 were active against all of the tested pathogens whereas compound 1 was active only against S. *flexneri* and S. aureus. No activity was noted for compound 5 (results not shown). Antimicrobial cut-off points have been defined by several authors to enable an understanding of the antimicrobial potential of pure compounds. Activity of a compound is classified as: significant activity (MIC < 10  $\mu$ g/mL), moderate activity ( $10 < MIC \le 100 \ \mu g/mL$ ), and low activity (MIC> 100  $\mu$ g/mL) (40,41). Accordingly, compound **3** had significant antibacterial activity against V. cholerae CO6, V. cholerae NB2, V. cholerae PC2, S. flexneri, and S. aureus while compound 6 had significant antibacterial activity against Shigella flexneri SDINT and Staphylococcus aureus ATCC 25923. The strains of V. cholerae NB2, PC2 (24,25) and Shigella flexneri (26) included in the present study were MDR clinical isolates and these were resistant to commonly used drugs such as ampicillin, streptomycin, tetracycline, nalidixic acid, furazolidone, and co-trimoxazole. However, most of the tested samples displayed antibacterial activity against these microbial strains, suggesting that their administration may represent an alternative treatment for V. cholerae, the cause of the dreadful disease cholera, and S. flexneri, the cause of shigellosis. Given the medical importance of the tested bacteria, the present results offer promise in terms of developing new antibacterials. The antibacterial activity of oleanolic acid, ellagic acid, and 2β-hydroxyoleanolic acid coincide with

Extracts/ compounds	Inhibition parameters	Vibrio cholerae SG24(1)	Vibrio cholerae CO6	Vibrio cholerae NB2	Vibrio cholerae PC2	Shigella flexneri SDINT	Staphylococcus aureus ATCC 25923
MeOH extract	MIC	512	256	256	256	128	64
	MBC	512	512	256	512	128	128
	MBC/MIC	1	2	1	2	1	2
	MIC	128	256	128	128	128	64
EtOAc extract	MBC	256	256	>512	256	128	128
	MBC/MIC	2	1	/	2	1	2
	MIC	256	256	128	256	128	64
	MBC	256	>512	256	256	128	128
<i>n</i> -BuOH extract	MBC/MIC	1	/	2	1	1	2
	MIC	> 256	>256	>256	>256	256	256
	MBC	/	/	/	/	>256	>256
1	MBC/MIC	/	/	/	/	/	/
	MIC	128	64	64	64	16	16
	MBC	128	64	64	128	32	16
2	MBC/MIC	1	1	1	2	2	1
	MIC	16	8	8	8	4	2
	MBC	16	8	8	8	4	2
3	MBC/MIC	1	1	1	1	1	1
	MIC	128	64	64	128	64	32
	MBC	> 256	128	64	256	64	32
4	MBC/MIC	/	2	1	2	1	1
	MIC	16	32	32	16	4	4
	MBC	32	32	32	16	8	8
6	MBC/MIC	2	1	1	1	2	2
	MIC	> 256	256	256	256	128	128
	MBC	/	>256	>256	>256	>256	128
7	MBC/MIC	/	/	/	/	/	1
	MIC	128	256	128	128	128	64
	MBC	> 512	256	256	256	128	64
8	MBC/MIC	/	1	2	2	1	1
	MIC	256	128	128	128	64	64
	MBC	> 256	256	128	256	128	64
9	MBC/MIC	/	2	1	2	2	1
	MIC	256	256	256	256	128	128
	MBC	> 256	>256	>256	>256	>256	128
10	MBC/MIC	/	/	/	/	/	1
	MIC	16	16	>512	>512	>512	4
	MBC	16	16	>512	>512	>512	4
Ampicillin	MBC/MIC	1	1	/	/	/	1
	MIC	0.5	2	0.5	0.5	16	2
	MBC	4	16	4	4	128	8
Tetracycline	MBC/MIC	8	8	8	8	8	4

Table 1. Antibacterial activit	v (MI	C and MBC	∑in ug/ı	ml) of extracts	s, isolated com	pounds, and	l reference antibacterials

/: not determined; MIC: Minimum Inhibitory Concentration; MBC Minimum Bactericidal Concentration.

previous findings (42,43). All of the compounds that were found to be active in the present study are members of the triterpenoid, flavonoid, and phenolic acid groups. Although triterpenoid, flavonoid, and phenolic acid compounds have been reported to possess antibacterial activity (39,44), no study has reported the activity of compounds **3** and **6-10** on the types of MDR pathogenic bacterial strains used in the present study.

The mechanism of action of terpenoids (1, 2, 5, and 7-10) is not fully understood, but it may involve membrane disruption by lipophilic compounds (45). Inhibition of the tested bacterial strains by phenolic acid (4) may be due to iron deprivation or hydrogen bounding with vital proteins such as microbial enzymes (46). The mechanism of action of flavonoids (3 and 6) is still to be studied; nevertheless, their activity is probably due

to their ability to form complexes with extracellular and soluble proteins and to form complexes with bacterial cell wall components. Moreover, lipophilic flavonoids may also disrupt microbial membranes (47).

#### 3.3. Antioxidant activity

The MeOH, EtOAc, and *n*-BuOH extracts and their isolated compounds were evaluated for their antioxidant activity using two *in vitro* models. The results were expressed as the gallic acid equivalent antioxidant capacity of tested samples (Figure 2) and as equivalent concentrations of test samples scavenging 50% of the DPPH radical (Figure 3). DPPH' and ABTS<sup>++</sup> radical scavenging activity were observed in all of the extracts. The MeOH and EtOAc extracts showed the most potent

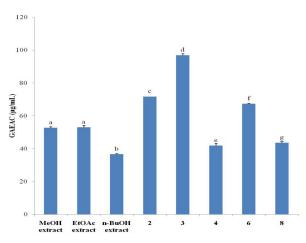


Figure 2. Gallic acid equivalent antioxidant capacity (GAEAC;  $\mu$ g/mL) of tested samples. Bars represent the mean  $\pm$  S.D. of three independent experiments carried out in triplicate. Letters a-e indicate significant differences between samples according to one-way ANOVA and the Waller Duncan test; p < 0.05. Compounds 1, 5, 7, 9, and 10 were not active (results not shown).

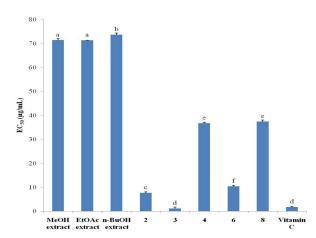


Figure 3. Equivalent concentrations of test samples scavenging 50% of the DPPH radical (EC<sub>50</sub>). Bars represent the mean  $\pm$  S.D. of three independent experiments carried out in triplicate. Letters a-f indicate significant differences between samples according to one-way ANOVA and the Waller Duncan test; p < 0.05. Compounds 1, 5, 7, 9, and 10 were not active (results not shown).

activity, followed by the n-BuOH extract (Figures 2 and 3). These results indicate the potential of the tested extracts to serve as a natural source of antioxidants with the potential to reduce oxidative stress and provide subsequent health benefits. The antioxidant capacity of the tested extracts may be due to the hydrogendonating ability of phenols and flavonoids present in those extracts. Similarly, previous studies have reported that phenolic compounds contribute significantly to the antioxidant activity of medicinal plants (39,48). The compounds that had the most potent DPPH' and ABTS<sup>++</sup> radical scavenging activity were compounds 2  $(EC_{50} = 7.66 \ \mu g/mL; \text{ GAEAC} = 71.64 \ \mu g/mL), 3 \ (EC_{50} = 7.66 \ \mu g/mL)$ = 1.09  $\mu$ g/mL; GAEAC= 96.88  $\mu$ g/mL), and 6 (EC<sub>50</sub> = 10.34  $\mu$ g/mL; GAEAC= 67.35  $\mu$ g/mL), while the other compounds (compounds 4 and 8) had moderate

antioxidant properties. Compounds 1, 5, 7, 9, and 10 were found to be inactive in both models. Compound 3 was the most potent antioxidant compound and its DPPH' radical scavenging activity was equal to that of vitamin C, which was used in the present study as reference antioxidant. This finding suggests that compound 3 is the best candidate to combat diseases associated with oxidative stress. This is very promising in terms of discovering antioxidants from plants. The antioxidant activity of compounds 2 and 4 agreed with previously reported findings (42,49). However, the present study is the first to document the antioxidant activity of the MeOH, EtOAc and *n*-BuOH extracts of *L. leptocarpa* as well as that of compounds 3, 6, and 8.

#### 4. Conclusion

Results indicated that MeOH and EtOAc extracts of *L. leptocarpa* as well as compounds **2**, **3**, and **6** possess the most potent antibacterial and antioxidant properties among the tested extracts and compounds. *L. leptocarpa* has the potential to be a natural source of products with health benefits, so it warrants further investigation.

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